



Extraction, Purification and Anti-Hyperlipidemic Activities of Total Flavonoids from Corn Silk

Jiandong Wu, Miao Ye and Zaigui Wang*

College of Life Sciences, Anhui Agriculture University, Anhui Hefei 230036, P.R. China

Jiandong Wu and Miao Ye contributed equally.

ABSTRACT

In this study, some important parameters were optimized for effective extraction and purification of total flavonoids from corn silk, and the anti-hyperlipidemic activities was analyzed in rabbits by feeding experiment. Firstly, the high extraction yield ($5.32 \text{ mg}\cdot\text{g}^{-1}$) of total flavonoids were obtained under the optimal extraction conditions, which are 80°C for 3 h with corn silk over 60% ethanol ratio of 1:30. Then, the AB-8 resin was selected for the purification of flavonoids due to the higher static adsorption and adsorption capacity than other resins. The optimum conditions for purifying flavonoids as follows: for adsorption, sample concentration was $2.542 \text{ mg}\cdot\text{mL}^{-1}$, pH was 3.7, processing volume was 4 BV; flow rate was $2 \text{ BV}\cdot\text{h}^{-1}$; temperature was 25°C . For desorption, elution solvent ethanol-water (95%, v/v) solution was 4.5 BV; flow rate was $2.0 \text{ BV}\cdot\text{h}^{-1}$. Flavonoids with content of 68.3% were obtained with a flavonoid recovery of 93.7%. The results showed that AB-8 resin revealed a good ability to purify total flavonoids. Lastly, feeding experiments results indicated that flavonoids had significant anti-hyperlipidemic activities to rabbits by reducing the content of serum total cholesterol (TC), triglycerides (TG), and low-density lipoprotein-cholesterol (LDL-C) and increasing the concentration of high-density lipoprotein-cholesterol (HDL-C).

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Authors' Contribution

JW and ZW designed the study, performed the experiments and wrote the manuscript. MY and JW interpreted the results.

Key words

Corn silk, Flavonoids, AB-8 resin, Purification, Anti-hyperlipidemic activities.

INTRODUCTION

Corn silk (dried cut stigmata of maize female flowers, *Zea mays L.*) is a well-known functional food and traditional Chinese herbal medicine. It consists of various bioactive constituents which has significant influence on human health. Corn silk had a long history of application for therapeutic remedy, including the antioxidant activities (Liu *et al.*, 2011), anti-diabetic activity on hyperglycemia rats (Zhao *et al.*, 2012) and the inhibitory effect on melanin production (Choi *et al.*, 2014). However, corn silk has been discarded as waste in most cases due to lack efficient method for extracting and purifying the bioactive constituents in China. Thus, how to explore an effective way for recycling the corn silk is a hotspot recently.

Flavonoids are important secondary metabolites widely exist in the plants. Some flavonoids or isoflavones were extracted from plants exhibited kinds of biological activities such as estrogenic, anti-oxidant activities, anti-osteoporosis, anti-diabetes and anti-carcinogenic effects (Wan *et al.*, 2014; Zhao *et al.*, 2012; Zhang *et al.*, 2007). Corn silk contains various compositions: proteins, vitamins, Ca, Mg and Na salts, oils, steroids, alkaloids, saponins, tannins, polysaccharide and flavonoids (Ebrahimzadeh *et al.*, 2008).

Flavonoids are one of the main constituents of corn silk and are considered to be an important active ingredient in corn silk. Previous studies showed that flavonoids from corn silk have many pharmacological effects, such as anti-oxidative (Liu *et al.*, 2011) and anti-fatigue activity (Hu *et al.*, 2010). However, little research about the anti-hyperlipidemic activities of corn silk was reported.

The preparative enrichment of active components from plant extracts is an important step in the manufacture of phytochemical-rich products. There are several methods, such as liquid-liquid extraction (Biazus *et al.*, 2007), ion exchange (Smith and Evens, 1995) and expanded bed adsorption (Biazus *et al.*, 2006), available in the literature for the enrichment and separation of active constituents. However, these methods are not particularly effective regarding reagents, energy consumption and labor intensiveness. Comparatively, macroporous resins become one of the most effective methods with purification process for the purification and recovery of plant secondary metabolites (Fu *et al.*, 2006; Kim *et al.*, 2007; Qin and Jin, 2014) due to high adsorption capacity with possible recovery of the adsorbed molecules, relatively low cost and simple procedure (Liu *et al.*, 2004).

Therefore, the extraction and purification conditions of total flavonoids were firstly optimized in this paper. Then the AB-8 resin was selected for the purification flavonoids due to the higher adsorption and desorption rate than other resin in the purification process. Through the purification of

* Corresponding author: wangzaigui2013@163.com
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AB-8 resin, the total flavonoids concentration was notable improved compared to the crude flavonoids. Lastly, the anti-hyperlipidemic activity of flavonoids was determined by the rabbits feeding experiment.

MATERIALS AND METHODS

Chemicals and materials

Analytical-grade methanol, ethanol, hydrochloric acid, sodium hydroxide and aluminium trichloride were purchased from the Guoyao biotech company. Standard samples of rutin were supplied by the National Institute of Control of Pharmaceutical and Biological Products (China). The macroporous resins (AB-8, ADS-8, NKA-2 and X-5) were purchased from Chemical Plant of Nankai University (Tianjin, China). Rabbits weighing 300-400 g were purchased from Experimental Animal Center, Anhui agricultural University, China. Corn silk was collected from the horticultural center of Anhui agricultural university (Hefei, China). Corn silk was dried for 24 h by using a hot air oven at 70°C and powdered and then was stored in a drying cabinet before the experiment.

Sample preparation and analytical methods

Extraction of flavonoids from corn silk was performed as following: the dried powder (5 g) was subjected to hot continuous extraction in Erlenmeyer flask using 50-65% ethanol at the material ratio of the weight (g) of corn silk to the volume of ethanol (mL) of 1:20 to 1:35. The resulting mixture was heated at the temperature of 60-90°C for 1-4 h and stirred constantly. Then the extract was filtrated by using the filter. The obtained filtrate was evaporated by using a rotary evaporator (RE-52A, Shanghai Yalong Co., Ltd., Shanghai, China) under vacuum at 60°C to get the crude flavonoids. The extraction solution was stored at 4°C until used.

The standard curve of flavonoids content was analyzed using rutin as the standard at 400 nm, using an ultraviolet (UV) spectrophotometer, and the standard curve was obtained, $C = 0.03040X + 0.00097$ with ($R^2 = 0.99986$). Here, C is the concentration (in units of mg/L) of flavonoids in the solution and A is the absorbance of the solution. The UV-2300 spectrophotometer (Shanghai Tianmei Co., Ltd., Shanghai, China) was used and wavelength was set at 400 nm.

Preparation of resin

Before use, the AB-8 macroporous resin was soaked in 95% ethanol for 24 h and washed several times using the deionized water. Then the resins were placed into a 1.2-cm inner diameter (id) glass column, and 2 BV (where BV represents the column bed volume) of 5% HCl and

the same quantity of NaOH (5%) were sequentially passed through the column at a flow rate of 2 BV·h⁻¹ to remove impurities. Finally, the column was washed with deionized water at the same flow rate until the effluent was chemically neutral.

Static adsorption and desorption tests

In the static adsorption experiment, 1 g of resin and 20 mL of crude flavonoids solution (2.542 mg·mL⁻¹) were introduced into an air-tight Erlenmeyer flask of 150 mL capacity, which was then shaken (100 r·min⁻¹) in a water-bath shaker at 25°C for 24 h to establish equilibrium. Then the concentration of flavonoids left in the solution was analyzed using spectrophotometer at 400 nm according to the standard curve.

The desorption process was conducted as follows: After adsorption equilibrium was reached, the resin was first washed by deionized water and then desorbed with 40 mL 95% ethanol solution. The flask was shaken (100 r·min⁻¹) in a water-bath shaker at 25°C for 24 h. Then the concentration of flavonoids left in the solution was analyzed using spectrophotometer at 400 nm.

Dynamic adsorption and desorption tests

Dynamic adsorption and desorption experiments were carried out in a glass column (1.2 × 40 cm) wet-packed with 4 g of the selected AB-8 resin. The bed volume (BV) of resin was 20 mL. The flow rate of sample (1-4 BV·h⁻¹) through the glass column, sample concentration (1-3 mg·mL⁻¹) and different pH values of crude flavonoids were optimized in the adsorption experiments at 25°C. Then the sample solution in a reasonable pH was pumped through the glass column at optimized flow rate by a peristaltic pump (ZT60-600, Baoding Lange Constant Pump Co., Ltd., Hebei, China). After adsorption equilibrium at the optimized condition, the adsorbate-laden column was successively eluted by deionized water and 95% ethanol at prescribed flow rate. The concentrations of flavonoids in the effluents (5mL/tube collected) were determined by spectrophotometer analysis.

Calculation of adsorption capacity, ratios of adsorption and desorption and recovery yield

$$\text{Adsorption capacity (Q)} = Q \frac{(C_0 - C_e)V_0}{W}$$

Where, Q is adsorption capacity at adsorption equilibrium (mg·g⁻¹ resin), C_0 and C_e are the initial and equilibrium concentrations of corn silk in the solutions, respectively (mg·mL⁻¹), V_0 is the volume of the initial sample solution (mL) and W is the weight of the dry resin (g).

$$\text{Adsorption ratio } (E_1 \%) = \frac{C_0 - C_1}{C_0} \times 100\%$$

Where, E_1 is the adsorption ratio (%), C_1 represents the concentration of the solute in the supernatant after 24 h ($\text{mg}\cdot\text{mL}^{-1}$) and C_0 is the same as above.

$$\text{Desorption ratio } (E_2 \%) = \frac{C_2}{C_0 - C_1} \times 100\%$$

Where, E_2 is the desorption ratio (%), C_2 is the concentration of flavonoids in the desorption solution ($\text{mg}\cdot\text{mL}^{-1}$) and C_0 and C_1 are the same as above.

$$\text{Recovery yield } (Y\%) = \frac{C_2 V_2}{(C_0 - C_a) V_p} \times 100$$

Where, Y is the recovery yield (%), C_a is the concentration of flavonoids in the effluent liquid ($\text{mg}\cdot\text{mL}^{-1}$), V_p is the processing volume of the sample solution (mL), V_2 is the volume of the desorption solution (mL) and C_0 and C_2 are the same as above.

Animals and grouping

Rabbits weighing 300-400 g were purchased for the feeding experiment. Rabbits were housed in plastic cages under controlled condition and fed on well-balanced diet (20% casein, 10% soybean oil, 5% cellulose, 4% mineral mix, 2% vitamin mix, 6% sucrose, and 53% cornstarch). Twenty rabbits were randomized into 2 groups (experiment group and control group) equally based on body weight and external characteristics after one week adoption. Rabbits of the experiment groups were fed on balanced diet supplemented with 0.2% of purified flavonoids and the control group were received the same volume of balanced diet for 21 consecutive days, respectively (Kim *et al.*, 2009; Weggemans and Trautwein, 2003).

Anti-hyperlipidemic activities

After 21 days, Blood samples of experiment group and control group were collected through heart puncture, allowed to coagulate and centrifuged to obtain serum for biochemical analysis. Then the concentration of TC, TG, LDL-C and HDL-C in the blood from rabbits of experiment group and control group were determined according to procedure described by Allian *et al.* (1974), Fossati and Prencipe (1982) and Jing *et al.* (2010). The data were analyzed with SPSS 10.0 software.

RESULTS AND DISCUSSION

Extraction of total flavonoids from corn silk

The ethanolic method was widely used for the extraction of total flavonoids from plants (Wan *et al.*,

2014; Li *et al.*, 2011; Feng *et al.*, 2012). In this processes, some important factors directly determine the extracting efficiency. So for maximizing the extraction yields of total flavonoids and minimizing the production cost, the different extraction conditions of total flavonoids from corn silk were analyzed. The yield of total flavonoids increased with increasing extraction time (up to about 3 h) and temperature (up to about 80°C) (Fig. 1A, B), with extraction temperature being more significant (the yields increased about 1.5 folds from 70°C to 80°C). Insufficient extraction duration yielded less flavonoids, and the yield can reached a maximum after 3-4 h of extraction (Fig. 1A). Furthermore, a higher extraction temperature of 80°C was preferred for maximizing the ratio of the yield of total flavonoids to the energy consumption (Fig. 1B). During extraction, the yield of total flavonoids from corn silk was also found to be associated with the material ratio of the weight of corn silk to the volume of ethanol (60%). A moderate ratio of 1:30 ($\text{g}\cdot\text{mL}^{-1}$) was preferred for its higher yield and lower cost requirement than the ratios of 1:25 and 1:35 (Fig. 1C). To reduce the cost, a relatively lower solvent, from 50% to 65%, were tested in experiment. The results showed that 60% ethanol can obtain higher yields than other concentration (Fig. 1D). It was probably due to the similar polarity between the 60% ethanol and the flavonoids (Rostagno *et al.*, 2003). Thus, 60% ethanol was selected as the solvent for extraction. In summary, the optimum extraction conditions were determined to be 80°C for 3 h with corn silk over 60% ethanol ratio of 1:30 to achieve the best extraction of total flavonoids. Under the optimum extraction conditions, about 5.32 mg of total flavonoids were separated from one gram of corn silk.

Adsorption and desorption capacities, desorption ratio of resins

Enrichment of active components from plant extraction is important step for the usage of active constituents. Due to its low cost, high efficiency and simple procedure, resin adsorption seems to be the most suitable method compare with the membrane filtration, ion exchange and expanded bed adsorption. Four macroporous resins with different properties (AB-8, ADS-8, NKA-2 and X-5) were tested through static adsorption and adsorption capacity, ratios of adsorption and desorption were used to evaluate their adsorbent efficiency. As shown in Table I, the adsorption and desorption ratios of total flavonoids on AB-8 resins were significantly higher than those of other resins, this result was consistent with previous study (Hu *et al.*, 2010; Zhao *et al.*, 2011). The adsorption capacities of the total flavonoids on AB-8 resins was 20.52 $\text{mg}\cdot\text{g}^{-1}$ dry resin with desorption ratio of 94.8%. Therefore, the AB-8 resin was selected for the purification experiments.

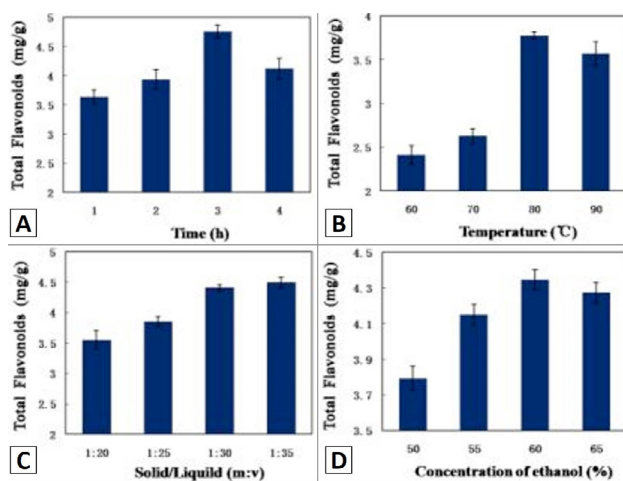


Fig. 1. Yield of total flavonoids from corn silk (mg/g) by extraction: A, for different durations with the material ratio (the weight of corn silk to the volume of 60% ethanol: g/mL) of 1:30 at 80°C; B, at different temperature with the material of 1:30 for 3 h; C, with different material ratio at 80°C for 3 h; D, for different ethanol concentration with material ratio of 1:30 for 3 h at 80°C. Experiments were repeated at least three times with similar results.

Table I.- Adsorption capacities, adsorption and desorption ratios of different resins.

| Varieties of resin | Adsorption capacity (mg/g) | Adsorption ratio (%) | Desorption ratio (%) |
|--------------------|----------------------------|----------------------|----------------------|
| AB-8 | 20.52 | 49.51 | 94.77 |
| ADS-8 | 15.86 | 38.17 | 81.12 |
| NKA-2 | 14.65 | 31.99 | 77.47 |
| X-5 | 10.26 | 14.82 | 75.41 |

Optimization of adsorption and desorption parameters for total flavonoids

To obtain the highest enrichment of total flavonoids with AB-8 resin, the effect of pH value of crude flavonoids, the flow rate, sample concentration and purification temperature on the adsorption rate were investigated in the flowing experiments. As shown in the Figure 2A, the highest adsorption capacity for the total flavonoids appeared at the pH values of 3.7, and then decreased with the increase of pH value. These results suggested that hydrogen bonding is critical to the adsorption process on AB-8 resin. The affinity between solutes and solutions is changed due to the pH influence the extent of ionization of solutes (Zhao *et al.*, 2011). Due to the dissociated of phenolic hydroxyl groups in the total flavonoids to H⁺ and their corresponding anions at higher pH value, thus resulting in the adsorption capacity reducing. Sample solutions

with different concentration of the total flavonoids (1-3 mg·mL⁻¹) were also prepared for purification. As can be seen from Figure 2B, with the increasing of concentration from 1 mg·mL⁻¹ to 2.5 mg·mL⁻¹, the desorption capacity significantly increased. When the concentration is up to 2.5 mg·mL⁻¹, the adsorption rate is at its maximum. Over the 2.5 mg·mL⁻¹, the adsorption rate is decreasing.

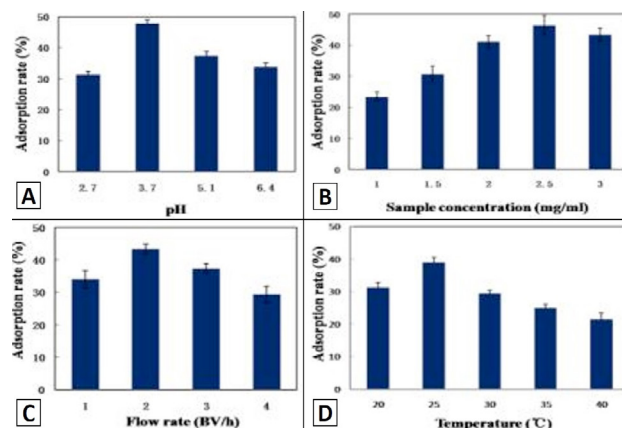


Fig. 2. The adsorption rate of macroporous resin AB-8 by purification: A, with different pH value of sample; B, at different sample concentration (mg/mL); C, for different flow rate (BV/h); D, at different temperature (°C). Experiments were repeated at least three times with similar results.

In addition, the effect of flow rate on dynamic adsorption capacity was tested. The results showed that the flow rate of 2 BV·h⁻¹ has larger adsorption capacity than flow rates of others (Fig. 2C). The reason for this is likely that the low flow rate was favorable to the diffusion in particle and liquid film of corn silk flavonoids molecules in the resin layer. Previous study showed that 25°C was the most suitable temperature for the adsorption of AB-8 resin (Wan *et al.*, 2014), similar results were found in our experiments (Fig. 2D).

The proper desorption solution was chosen according to the polarity of resins and the flavonoids solubility in the desorption solution. The desorption rate was carried out with different concentrations ethanol solution at the flow rate of 2 BV·h⁻¹. At the 30% ethanol, the total flavonoids were hardly desorbed. When the ethanol concentration was over 30%, the desorption rate increased sharply and reached 93.7% with the 95% ethanol (Fig. 3). Although the desorption rate with 100% ethanol was slightly higher than that with 95% ethanol, yet the 95% ethanol was selected for the later dynamic desorption curve analysis due to the lower cost.

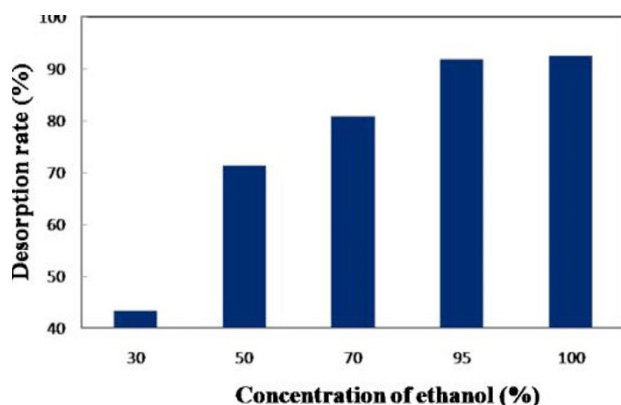


Fig. 3. Effect of different ethanol-water solutions on the desorption ratio of total flavonoids on AB-8 resin.

Dynamic breakthrough and desorption curve on AB-8 resin

The solute leakage from the column at breakthrough point was due to the resin in the column approached to be saturated. The breakthrough point was defined as 10% of the ratio of the exit to the inlet solution concentration (Hu *et al.*, 2010). Therefore, construction of the leakage curve is important to calculate the quantity of resins, the processing volume of sample solution and the proper sample flow rate. The dynamic leakage curves on AB-8 resin were obtained according to the volume of effluent liquid and the tested constituents' concentration. As shown in Figure 4A, flavonoids in solution was almost completely adsorbed by the AB-8 resin before 4 BV (breakthrough point), and then the concentration of flavonoids in leak solution increased rapidly until it reached a steady plateau in 7 BV. Therefore, the 4 BV of sample solution on AB-8 resin was selected for dynamic adsorption experiments.

According to the results of breakthrough volume determined above, the 4BV sample solution was fed on the column packed with 4 g AB-8 resin. After adsorption equilibrium the resin was first flushed with 4 BV of distilled water for removing the high polar components in the crude extraction of corn silk, such as polysaccharides and amino acids. And then the adsorbent was eluted with 95% ethanol aqueous. The dynamic desorption curves on AB-8 resin were obtained based on the volume of effluent and the concentration of solute therein. The flow rate in this test was 2 BV·h⁻¹. As can be seen in Figure 4B, the concentration of flavonoids in elution solution was increasing with the increasing of the volume of elute, reaching a maximum value at 2 BV and then decreased. Adsorption solution of 4.5 BV could completely desorbed flavonoids from AB-8 resin. In summary, the optimization of the desorption conditions: elution solvent ethanol-water (95%, v/v) solution was 4.5 BV; flow rate was 2 BVh⁻¹.

Flavonoids with content of 68.3% was obtained with a flavonoids recovery of 93.7% in the purification process.

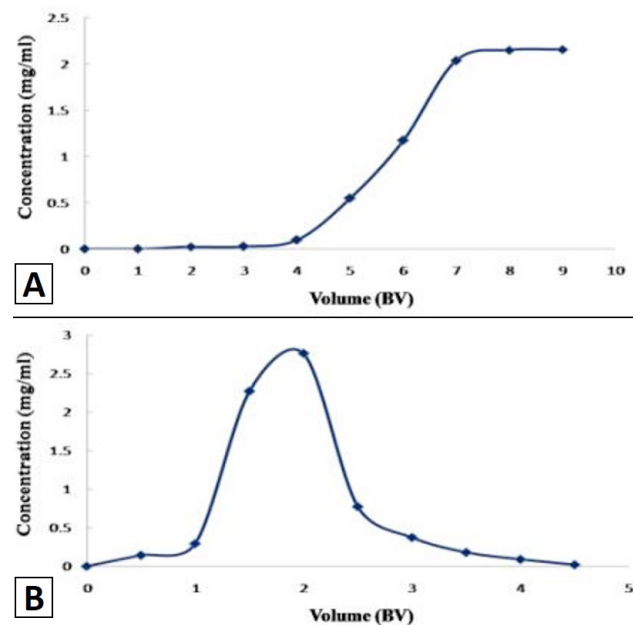


Fig. 4. Dynamic breakthrough (A) and dynamic adsorption (B) curves of total flavonoids on column packed with AB-8 rein. Experimental condition: temperature, 25°C; flow rate, 2 BV/h; concentration of total flavonoids, 2.5 mg/mL.

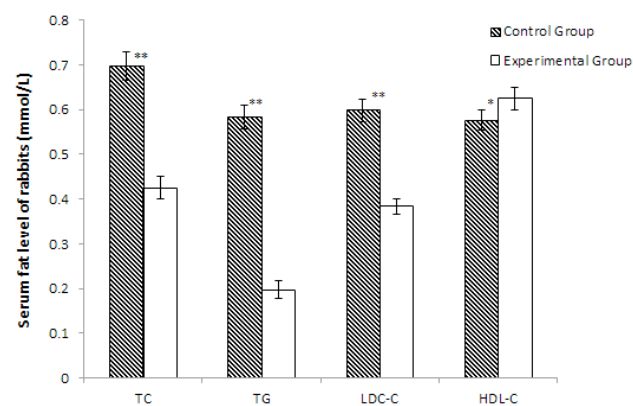


Fig. 5. Effects of total flavonoids on the serum flat level in rabbits. Data are expressed as mean \pm SEM, each group contains ten mice (n=10), *, p<0.05; **, p<0.01. TC, total cholesterol; TG, triglycerides; LDC-C, low-density lipoprotein-cholesterol; HDL-C, high-density lipoprotein-cholesterol.

Anti-hyperlipidemic activities

The concentration of serum TC, TG, LDL-C and HDL-C were important parameters to serum lipids level (Bibi *et al.*, 2016; Jung *et al.*, 2014; Hamza *et al.*, 2013;

Jing et al., 2010). Weggemans and Trautwein (2003) also reported that flavonoids intake decreased LDL-C and increased HDL-C that may enhance removal of cholesterol from peripheral tissue to liver for catabolism and excretion. To determine the anti-hyperlipidemic activities of total flavonoids from corn silk to rabbits. The content of TC, TG, LDL-C and HDL-C of rabbit serum lipids from experiment group and control group were detected. As previous study, the content of TC, TG and LDL-C of experimental group rabbits had significantly reduction compared to control group after three weeks feeding (Fig. 5), and the decreasing rate were 60.9%, 194.2% and 64.9%, respectively. However, the concentration of HDL-C in serum of feeding experimental group rabbits was higher than the control group. These results suggested that total flavonoids of corn silk have notable anti-hyperlipidemic activity.

CONCLUSIONS

In this study, the total flavonoids from the corn silk was extracted with higher content under the optimum extraction conditions were determined to be 80°C for 3 h with corn silk over 60% ethanol ratio of 1:30. The AB-8 resin was selected for the purification of crude flavonoids because of the higher absorption and desorption rate. In this process, the impact factor of desorption capacity and desorption rate were optimized. The results were as follows: for adsorption, sample concentration was 2.542 mg·mL⁻¹, pH was 3.7, processing volume was 4 BV; flow rate was 2 BV·h⁻¹; temperature was 25°C; for desorption, elution solvent ethanol-water (95%, v/v) solution was 4.5 BV; flow rate was 2.0 BV·h⁻¹; Flavonoids with content of 68.3% was obtained with a flavonoid recovery of 93.7% after purification. These results showed that the flavonoids were purified successfully with the higher content. In addition, flavonoids from corn silk had notable anti-hyperlipidemic activities by reducing the concentration of serum TC, TG and LDL-C, and increasing the concentration of HDL-C.

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Statement of conflict of interest

Authors have declared no conflict of interest.

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